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Hydrogen bonding in zirconium sulfate tetrahydrate.* By DAVID H. TEMPLETON, *Department of Chemistry and Lawrence Radiation Laboratory, University of California, Berkeley, California, U.S.A.*

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In reporting the structure of zirconium sulfate tetrahydrate, Singer & Cromer (1959) suggested a configuration for the hydrogen bonds which placed a hydrogen atom between two oxygen atoms in the same coordination polyhedron of zirconium. It is expected that a water molecule coordinated to a cation will have its hydrogen atoms on the side away from the cation. In several hydrated sulfates, e.g., $\text{NiSO}_4 \cdot 6\text{H}_2\text{O}$ (Beevers & Lipson, 1932), $\text{NiSO}_4 \cdot 7\text{H}_2\text{O}$ (Beevers & Schwartz, 1935), $\text{CuSO}_4 \cdot 5\text{H}_2\text{O}$ (Beevers & Lipson, 1934), and $\text{KAl}(\text{SO}_4)_2 \cdot 12\text{H}_2\text{O}$ (Lipson & Beevers, 1935), there are just enough short oxygen-oxygen distances to account for all of the hydrogen bonds, if one excludes from consideration the short distances between oxygen atoms in the same coordination polyhedron or in the same sulfate group.

An examination of the structure of $\text{Zr}(\text{SO}_4)_2 \cdot 4\text{H}_2\text{O}$, as reported by Singer & Cromer (1959), reveals a more plausible assignment. Each water oxygen, O_3 , has four close neighbors in the same square antiprism, O_2 at 2.53, 2.62, and 2.86 Å and O_3 at 2.72 Å. It has three

other neighbors, O_1 at 2.69, 2.75, and 2.93 Å. It is reasonable to assign the hydrogen bonds to the 2.69 and 2.75 Å distances. The angle between these two bonds is 88° , and the bisector of this angle, within experimental error, is 180° from the line from O_3 to Zr.

References

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Notes and News

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The Sixth Annual Conference on Magnetism and Magnetic Materials will be held in New York City, 14-17 November 1960, at the New Yorker Hotel. This conference is sponsored jointly by the American Institute of Electrical Engineers and the American Institute of Physics, in cooperation with the Office of Naval Rese-

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Authors should submit titles and abstracts of proposed papers by 26 August to A. M. Clogston or R. C. Fletcher, Programme Chairmen, Bell Telephone Laboratories, Murray Hill, New Jersey. Further conference details can be obtained from the Local Chairman, L. R. Bickford, Jr., IBM Research Center, Yorktown Heights, New York.